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_____ 1. The principal quantum number of an electron is 4. What are the

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Date Chapter Test A, continued _____ 8. What is the frequency of a photon whose energy is 3.4×10^{-19} J? ($h = 6.626 \times 10^{-34}$ J·s) a. 8.8×10^{26} Hz b. 5.1×10^{14} Hz c. 1.9×10^{-15} Hz d. 2.3×10^{-52} Hz _____ 9. When electromagnetic radiation strikes the surface of a metal, electrons are ejected from the metal's ...

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(b) particle nature. (d) wave nature. 6. c According to Bohr, which is the point in the figure below where electrons cannot reside? (a) point A (c) point C (b) point B (d) point D 7. c According to the quantum theory, point D in the above figure represents (a) the fixed position of an electron. (b) the farthest position from the nucleus that an ...

4 Arrangement of Electrons in Atoms

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_____ _____ refers to the arrangement of conditions in a study so that meaningful comparisons of the effect of the IV can be made

1. intervention
2. experimental design
3. single-subject design
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